

APPENDIX III

Performance Characteristics of Revised Plans.—The "2-step" plan described in *Recent Revisions* can be defined as follows:

Compute the arithmetic mean, \bar{x} , of the 10 sample values and accept the lot if the following conditions hold: (A) $0.95 \mu_0 < \bar{x} < 1.05 \mu_0$, and (B) $|x_i - \bar{x}| < 0.07 \bar{x}$ for at least 8 of the 10 samples values, x_i .

In the notation of the previous section, the probability that condition (A) holds is simply $P_A(\mu, \sigma)$, as given in Eq. 1. The probability, $P_B(\mu, \sigma)$ that condition (B) holds can be approximated by assuming that $x_i - \bar{x}$ and $x_j - \bar{x}$ are statistically independent when $i \neq j$. Letting $h = P_r[|x_i - \bar{x}| < 0.07 \bar{x}]$ we obtain for P_B the binomial type expression

$$P_B(\mu, \sigma) = \sum_{r=8}^{10} \binom{10}{r} h^r (1-h)^{10-r}$$

where $h = 2F(0.0738 \mu/\sigma) - 1$.

The value in Table I corresponding to a particular normal distribution was obtained by first computing μ , σ , $P_A(\mu, \sigma)$, $P_B(\mu, \sigma)$, and then approximating the probability of acceptance for the revised plan as $P'_A(\mu, \sigma) = P_A(\mu, \sigma) P_B(\mu, \sigma)$.

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Notes

The Occurrence of *isoPelletierine* in *Withania somnifera*

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INVESTIGATION of the root alkaloids of *Withania somnifera* Dunal has resulted in the delineation of the alkaloid complex by paper partition chromatography (1), the isolation of tropine and pseudotropine (2), and the isolation and characterization of a new alkaloid, anaferine (3). Compound VII of the chromatogram, cited above, is *dl-isopelletierine*. The occurrence of this alkaloid in the *Solanaceae* was first reported (4) for the leaves of *Duboisia myoporioides* R. Br.; the previously acknowledged occurrence being the root of *Punica granatum* L., family *Punicaceae*. This finding amplifies the known biochemical heterogeneity of the *Solanaceae* and extends a lysine related alkaloid to yet another genus in the plant kingdom.

EXPERIMENTAL¹

The alkaloid was isolated from an ethanol extract of the defatted granulated root (81 Kg.). The concentrated extract diluted with water and adjusted to pH 4.7 was adsorbed on a column of Amberlite IRC-50-Na resin and was eluted by a gradient acid-buffer method. The fraction containing *isopelletierine* hydrochloride was further purified by chroma-

tography on acid alumina (Woelm, grade 1) using ethyl acetate as the eluent. The alkaloid salt (5.67 Gm.) was recrystallized from ethyl acetate.

isoPelletierine Hydrochloride.—The infrared spectrum was identical to that of a synthetic sample² and a natural sample;³ m.p. 145°, undepressed in admixture with each of the above samples; $[\alpha]_D^{25} = 0.00$ (0.21% in ethanol).

Anal.—Calcd. for $C_8H_{16}ClN$: C, 54.07; H, 9.07; Cl, 19.95; N, 7.88. Found: C, 54.38; H, 9.59; Cl, 20.21; N, 7.67.

isoPelletierine Picrate.—The compound was prepared from the hydrochloride and was crystallized from ethanol; m.p. 148.5 to 149.5°, reported 149–150°. The melting point of a mixture with the picrate of an authentic sample² was undepressed.

isoPelletierine 2,4-Dinitrophenylhydrazone Hydrochloride.—The compound was prepared from the hydrochloride and was crystallized from ethanol-ethyl acetate; m.p. 240–241.5° (decompn.), lit. 242°.

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² Supplied by Prof. J. B. Wibaut and Prof. H. O. Huisman, Holland.

³ Supplied by Dr. P. I. Mortimer and Dr. J. W. Clark-Lewis, Australia.

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¹ Analysis by Geller Microanalytical Laboratories, Bardonia, N. Y. All melting points are corrected. Infrared spectra were determined in KBr pellets, using a Perkin-Elmer model 21 spectrophotometer.